Cambridge
O Level

Cambridge International Examinations
Cambridge Ordinary Level

CANDIDATE NAME

CENTRE NUMBER

CANDIDATE NUMBER

1248049088

BIOLOGY **5090/32**

Paper 3 Practical Test **May/June 2016**

1 hour 15 minutes

Candidates answer on the Question Paper.

Additional Materials: As specified in the Confidential Instructions.

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.
Write in dark blue or black pen.
You may use an HB pencil for any diagrams or graphs.
Do not use staples, paper clips, glue or correction fluid.
DO **NOT** WRITE IN ANY BARCODES.

Answer **all** questions.
Write your answers in the spaces provided on the Question Paper.

Electronic calculators may be used.
You may lose marks if you do not show your working or if you do not use appropriate units.

At the end of the examination, fasten all your work securely together.
The number of marks is given in brackets [ ] at the end of each question or part question.

For Examiner's Use	
1	
2	
3	
Total	

This document consists of **11** printed pages and **1** blank page.

© UCLES 2016

CAMBRIDGE International Examinations

https://xtremepape.rs/

In order to plan the best use of your time, read through all the questions on this paper carefully before starting work.

1 (a) Dehydrogenase is an enzyme found in cells such as yeast.
Methylene blue is an indicator that can be used to show the activity of dehydrogenase.

In the presence of active dehydrogenase, methylene blue changes colour from blue to colourless.

You are required to carry out an experiment to investigate the effect of temperature on the activity of dehydrogenase in yeast cells.

You have been provided with two yeast suspensions, **A** and **B**. Yeast suspension **B** has been heated to boiling and then cooled.

- Prepare a water bath by half filling a beaker with water and adjusting the temperature to 25 °C.
- Label one test-tube **A** and one test-tube **B**.
- Put 5 cm^3 of yeast suspension **A** into the test-tube labelled **A**.
- Put 5 cm^3 of yeast suspension **B** into the test-tube labelled **B**.
- Stand both test-tubes in the beaker of water kept at 25 °C. Leave the test-tubes in the water for 5 minutes.
- After 5 minutes, add 1 cm^3 of methylene blue solution to each test-tube, mix carefully and leave the test-tubes in the water.
- Observe the colour of the contents of the test-tubes every minute for a period of 5 minutes, while keeping the water temperature at 25 °C.
- Repeat this procedure using fresh yeast suspension and a beaker of water kept at 35 °C.

(i) Record all your observations in Table 1.1.

Table 1.1

temperature of water /°C	test-tube	observations at				
		1 min	2 mins	3 mins	4 mins	5 mins
25	A					
	B					
35	A					
	B					

[4]

© UCLES 2016

https://xtremepape.rs/

(ii) Explain how you were able to check the temperature of the water in the beaker.

.....

.....[1]

(iii) Explain how you kept the temperature of the water in the beaker at 25 °C and at 35 °C.

.....

.....[1]

(iv) Explain why the test-tubes were left in the beaker of water for 5 minutes before adding methylene blue solution.

.....

.....[1]

(v) Suggest an explanation for your observations.

.....

.....

.....

.....

.....

.....

.....[3]

https://xtremepape.rs/

(b) Giving full experimental details, describe an experiment you could carry out to investigate the effect of pH on the activity of dehydrogenase, using yeast suspension and methylene blue solution.

.....

.....

.....

.....

.....

.....

.....

.....

.....

.....

.....

.....

.....

.....[5]

(c) Yeast is used for the production of alcohol in the brewing industry.

In an investigation, the production of alcohol by yeast was recorded every 5 hours for 25 hours.

The results are shown in Table 1.2.

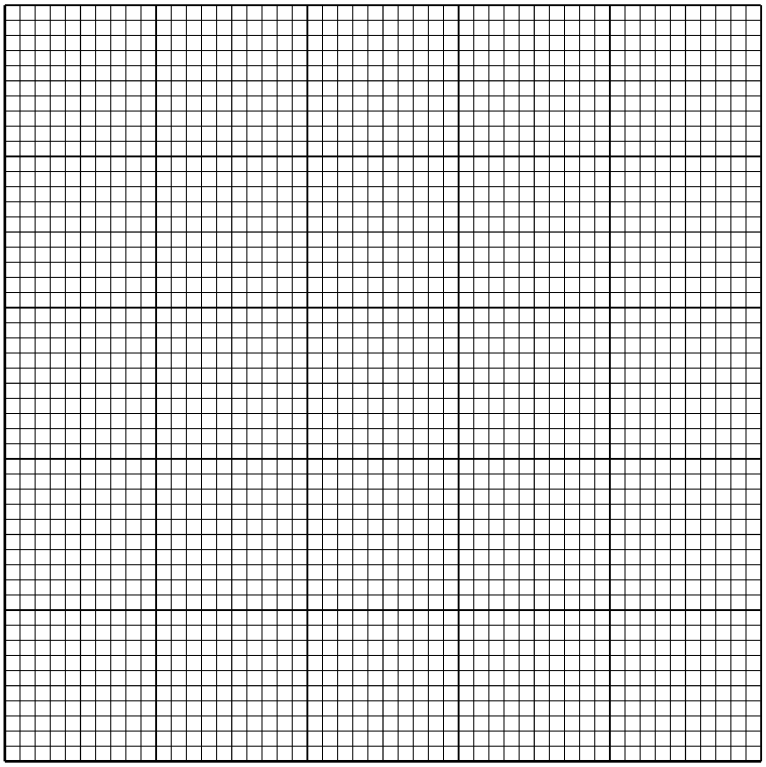
Table 1.2

time /hr	alcohol concentration /g per dm^3
0	0.0
5	3.7
10	5.6
15	6.8
20	7.5
25	7.8

© UCLES 2016

https://xtremepape.rs/

(i) Construct a line graph of the data in Table 1.2 on the grid below.
Draw a smooth curve through your points.



[4]

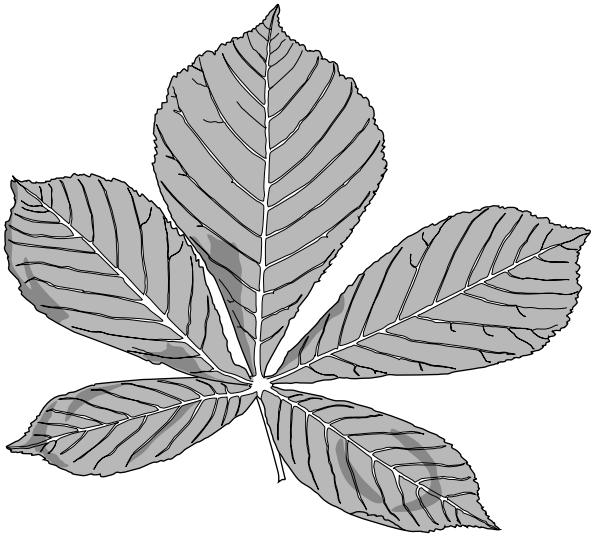
(ii) Use your graph to find the concentration of alcohol after 12 hours.

concentration = ... [2]

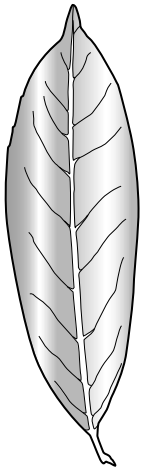
[Total: 21]

https://xtremepape.rs/

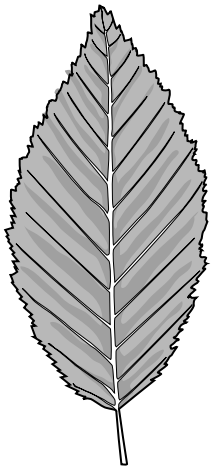
2 Fig. 2.1 shows leaves from four different trees: horse chestnut, laurel, hornbeam and oak.



horse chestnut



laurel



hornbeam

oak

Fig. 2.1

© UCLES 2016

https://xtremepape.rs/

(a) In the space below, make a large drawing of the horse chestnut leaf.
You do not need to label your drawing.

[5]

(b) There are several differences between these leaves, which can be used to identify each leaf.

Complete Table 2.1 by describing the overall shape and the edge (margin) of the laurel leaf and the oak leaf.

The shape and edge of the hornbeam have been described for you.

Table 2.1

feature	tree		
	hornbeam	**laurel**	**oak**
shape	oval		
edge (margin)	serrated		

[4]

https://xtremepape.rs/

(c) (i) The actual maximum width of the laurel leaf is 40 mm. Measure and record the maximum width of this leaf in Fig. 2.1.

Draw a line on Fig. 2.1 to show where you have taken this measurement.

maximum width = ...[2]

(ii) Calculate the magnification of the laurel leaf in Fig. 2.1.

Show your working.

magnification × .....[2]

[Total: 13]

https://xtremepape.rs/

Question 3 begins on page 10.

https://xtremepape.rs/

3 An investigation was carried out to study the effect of physical activity on breathing.

The breathing rate (number of breaths per minute) of a student was measured at rest. The student then cycled at a speed of 10 km per hour for 2 minutes and their breathing rate was measured immediately after.

They then rested for five minutes before cycling at 15 km per hour for 2 minutes. Their breathing rate was measured again.

This investigation was repeated for cycling speeds of 20 km per hour and 25 km per hour.

The results are shown in Table 3.1.

Table 3.1

cycling speed /km per hour	breathing rate /number of breaths per minute
rest	12
10	14
15	17
20	20
25	27

(a) (i) Using the information in Table 3.1, state the general relationship between cycling speed and breathing rate.

.....

.....[1]

(ii) Suggest an explanation for this relationship.

.....

.....

.....

.....

.....

.....

.....[3]

© UCLES 2016

https://xtremepape.rs/

(b) The volume of air breathed in and out by the student also changed during this investigation, as shown in Table 3.2.

Table 3.2

cycling speed / km per hour	volume of air breathed in and out in each breath / cm^3
rest	600
25	3000

The **minute volume** is found by multiplying the volume of air breathed in and out in each breath by the breathing rate.

Using the information in Table 3.1 and in Table 3.2, calculate the minute volume after cycling at 25 km per hour.

Show your working.

minute volume = ... cm^3 [2]

[Total: 6]

© UCLES 2016

https://xtremepape.rs/

BLANK PAGE

Permission to reproduce items where third-party owned material protected by copyright is included has been sought and cleared where possible. Every reasonable effort has been made by the publisher (UCLES) to trace copyright holders, but if any items requiring clearance have unwittingly been included, the publisher will be pleased to make amends at the earliest possible opportunity.

To avoid the issue of disclosure of answer-related information to candidates, all copyright acknowledgements are reproduced online in the Cambridge International Examinations Copyright Acknowledgements Booklet. This is produced for each series of examinations and is freely available to download at www.cie.org.uk after the live examination series.

Cambridge International Examinations is part of the Cambridge Assessment Group. Cambridge Assessment is the brand name of University of Cambridge Local Examinations Syndicate (UCLES), which is itself a department of the University of Cambridge.

© UCLES 2016

https://xtremepape.rs/